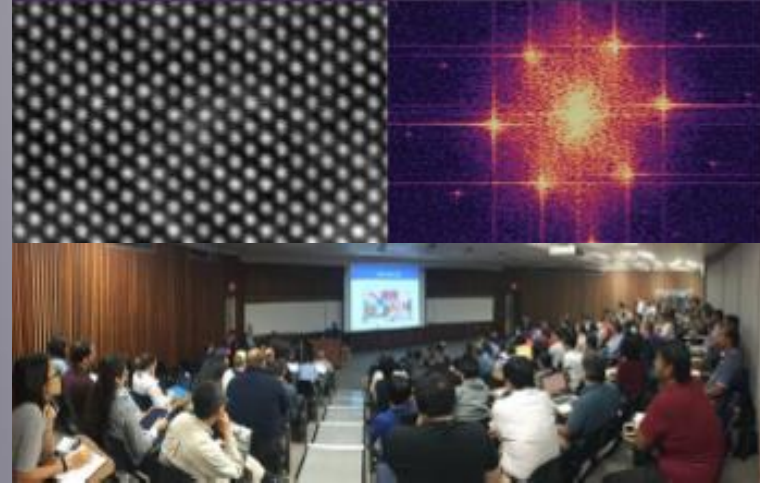


June 4 - 5, 2019

AMC

Advanced Materials
Characterization Workshop



Secondary Ion Mass Spectrometry

Timothy P. Spila, Ph.D.

Materials Research Laboratory
University of Illinois



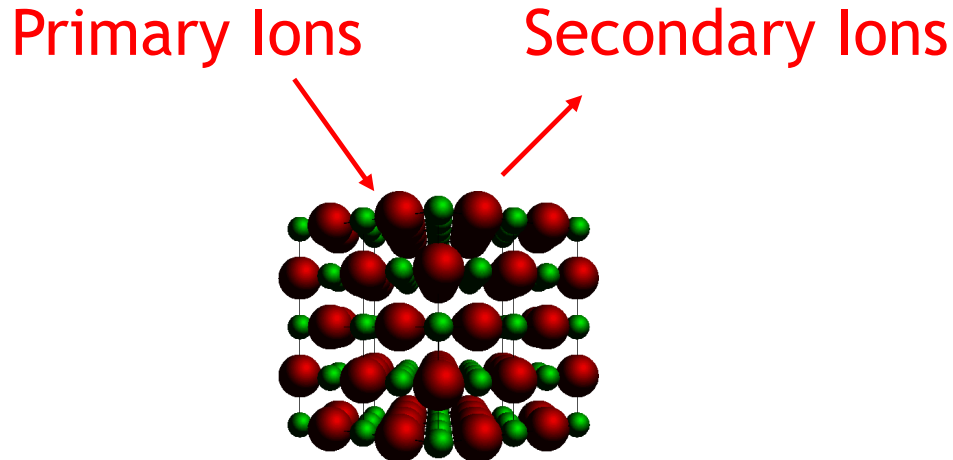
amc.mrl.illinois.edu





Secondary Ion Mass Spectrometry

SIMS is an analytical technique based on the measurement of the mass of ions ejected from a solid surface after the surface has been bombarded with high energy (1-25 keV) primary ions.





Technique Comparison

	AES	XPS	D-SIMS	TOF-SIMS
Probe Beam	Electrons	Photons	Ions	Ions
Analysis Beam	Electrons	Electrons	Ions	Ions
Spatial Resolution	8 nm	9 μm	2 μm	0.1 μm
Sampling Depth	0.5 – 7.5 nm	0.5 – 7.5 nm	0.1 – 1 nm	0.1 – 1 nm
Detection Limits	0.1 – 5 atom %	0.01 – 0.1 atom %	1 ppm*	1 ppm*
Quantification	Good Semi-quantitative	Excellent Semi-quantitative	Challenging Large matrix effects	Challenging Large matrix effects
Information Content	Elemental	Elemental Chemical bonding	Elemental	Elemental Molecular
Insulator Analysis	Challenging	Excellent**	Good**	Excellent**
Organic Analysis	Electron beam damages organics	Excellent	DC ion beam damages organics	Excellent in static mode
Depth Profiling	Excellent for small areas	Excellent for insulating materials	Excellent for speed and sensitivity	Excellent for sensitivity

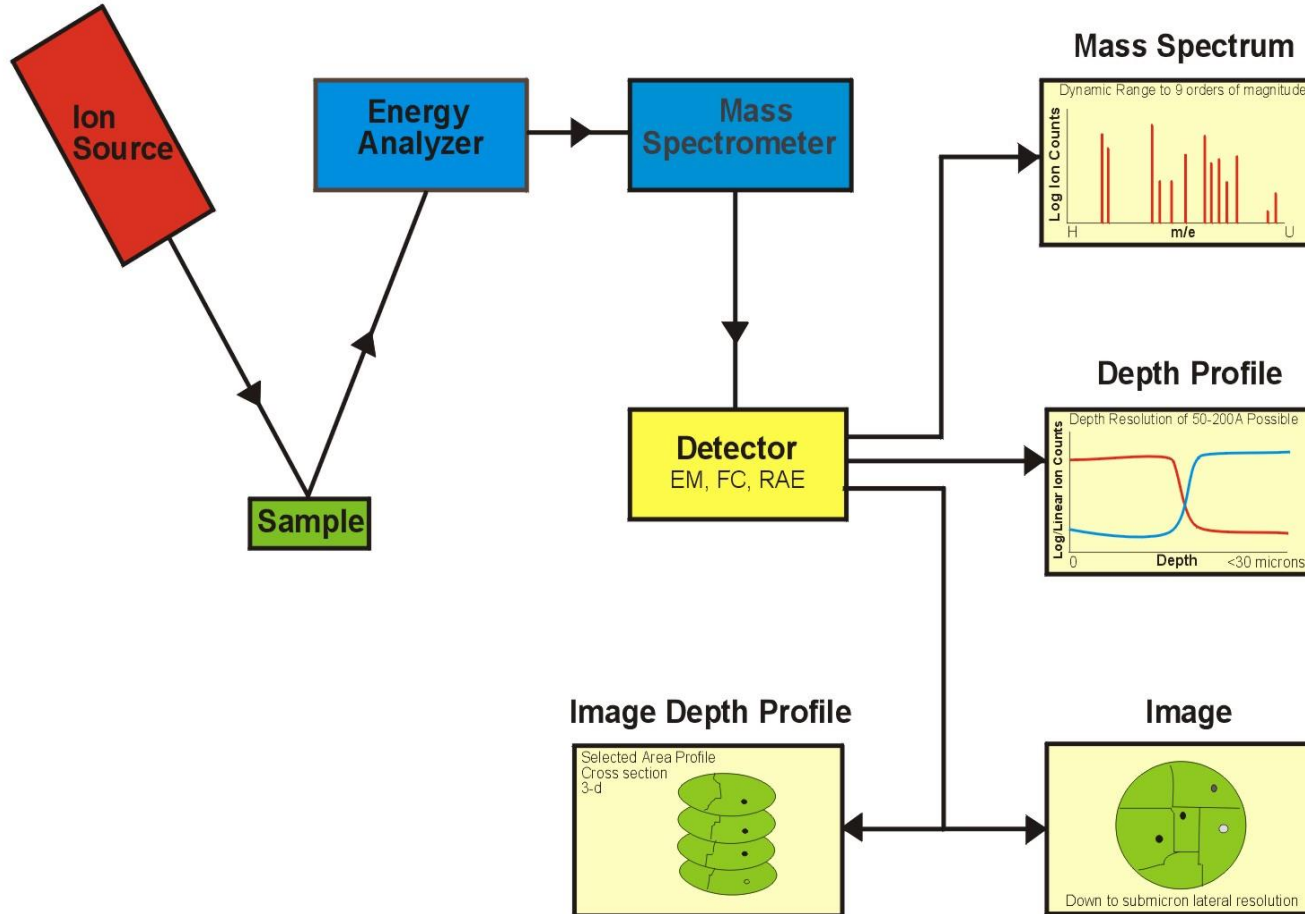
* 1 ppm sensitivity is achieved by consuming the sample surface

** requires effective charge neutralization apparatus



Block Diagram of SIMS Technique

2-20 keV Ar or other inert gases,
Cs, O, N, or Ga

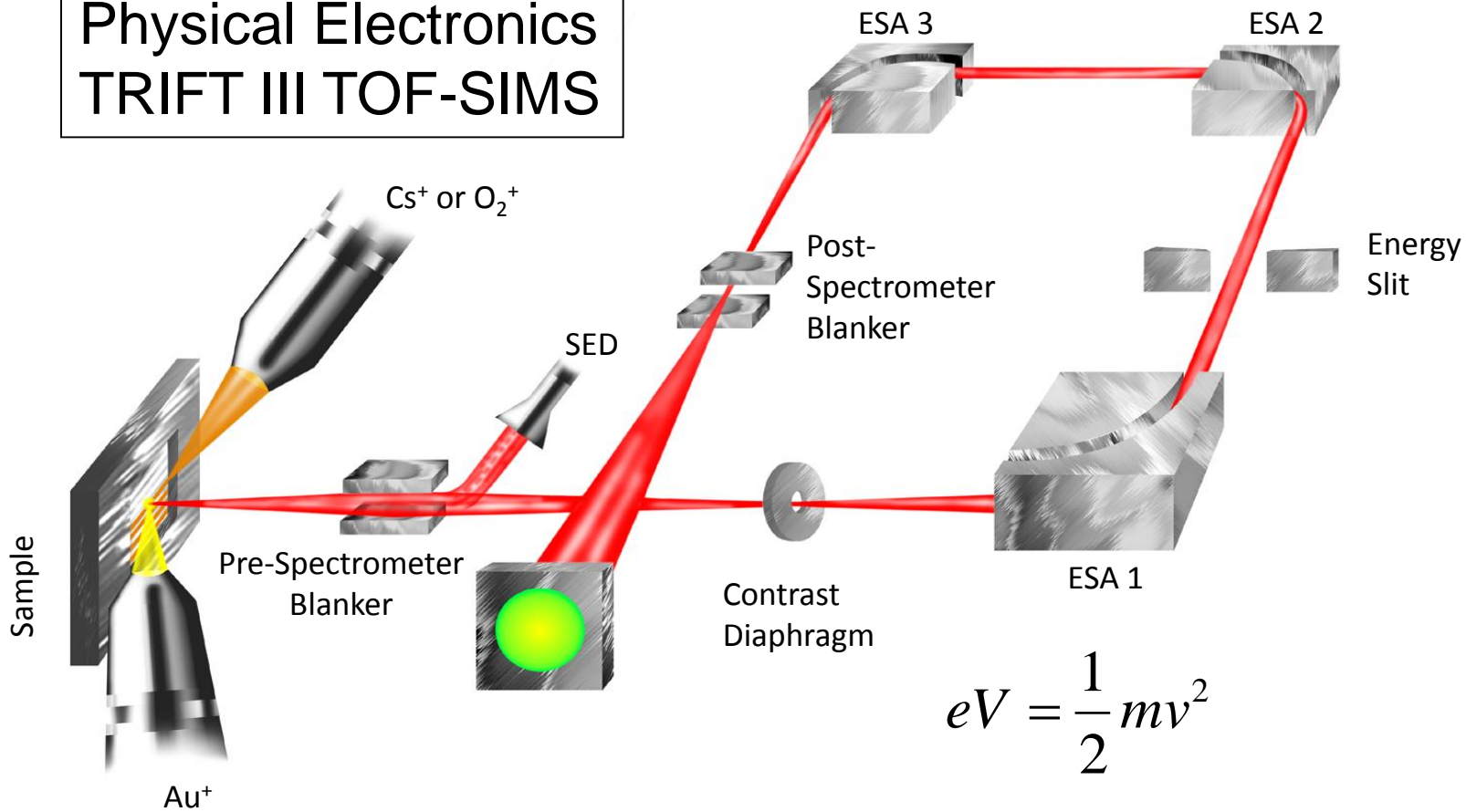


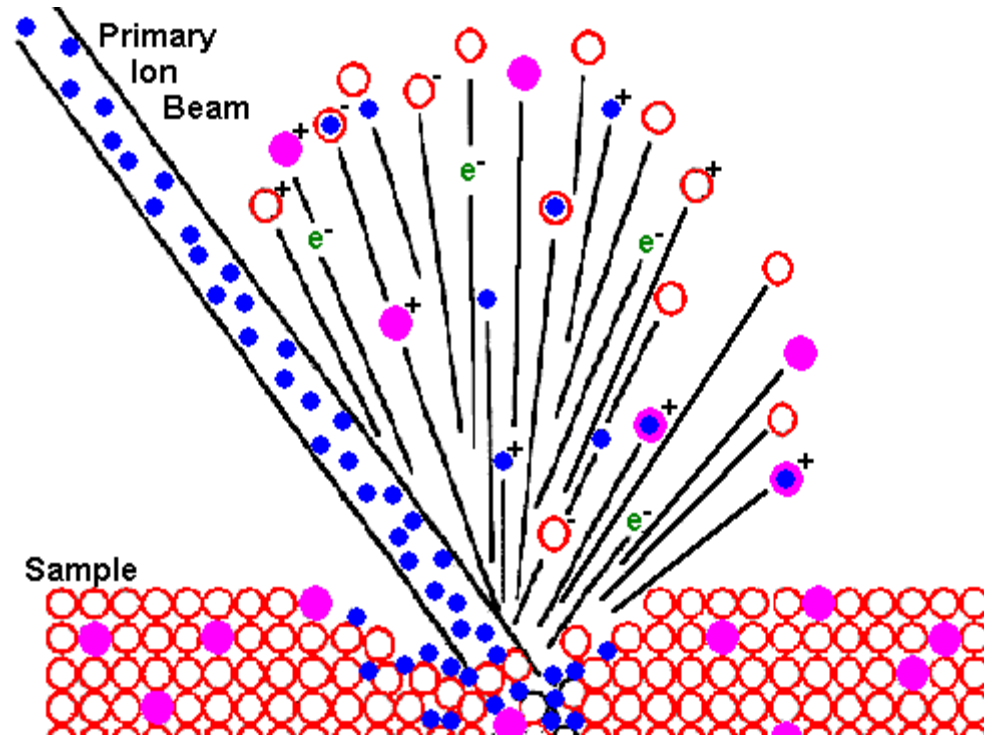
Adapted from Wilson, Stevie, and Magee, p. I-8.



Time of Flight Mass Spectrometer

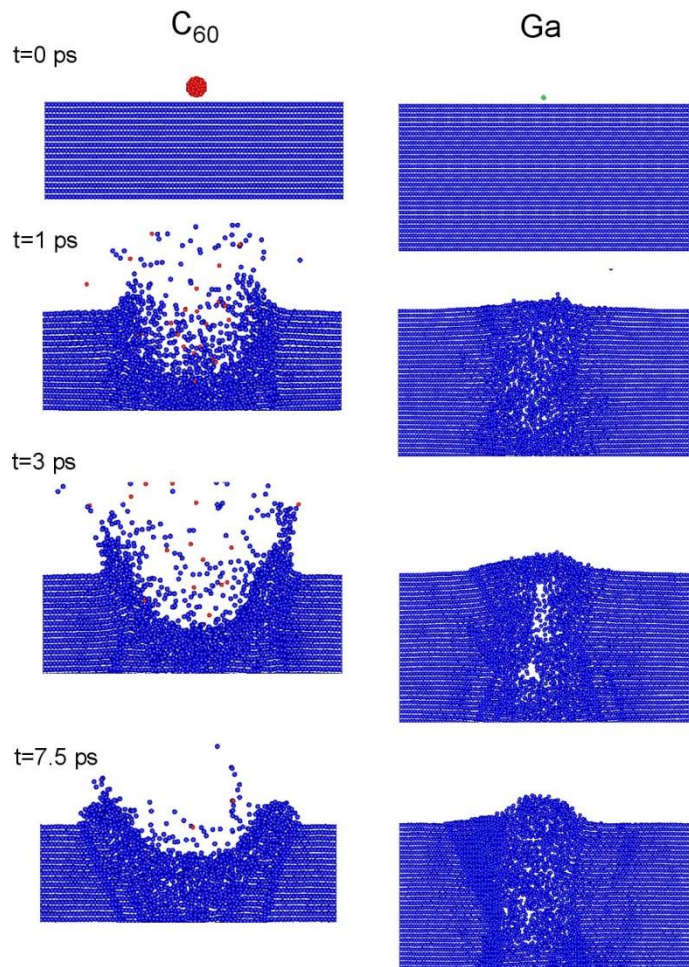
Physical Electronics
TRIFT III TOF-SIMS





Sputtered species include:

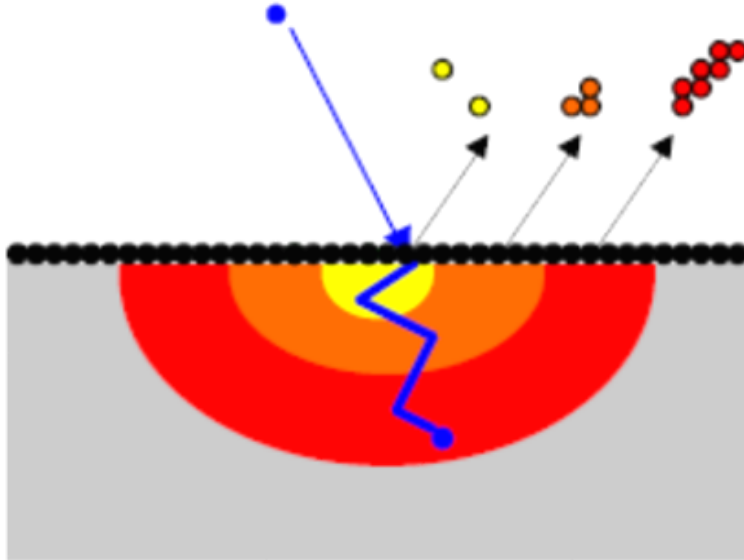
- **Monoatomic and polyatomic particles of sample material (positive, negative or neutral)**
- **Resputtered primary species (positive, negative or neutral)**
- **Electrons**
- **Photons**



Enhancement of Sputtering Yields due to C₆₀ vs. Ga Bombardment of Ag{111} as Explored by Molecular Dynamics Simulations, Z. Postawa, B. Czerwinski, M. Szewczyk, E. J. Smiley, N. Winograd and B. J. Garrison, *Anal. Chem.*, **75**, 4402-4407 (2003).

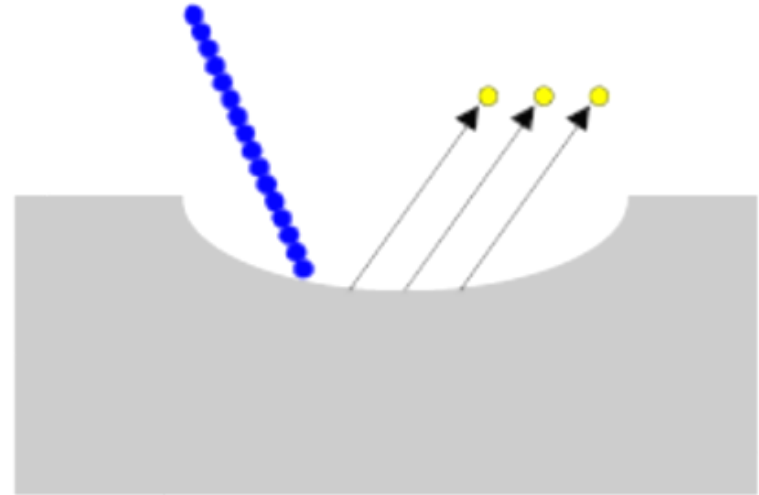
Animations downloaded from <http://galilei.chem.psu.edu/sputtering-animations.html>.

Static SIMS



- Ultra surface analysis
- Elemental or molecular analysis
- Analysis complete before significant fraction of molecules destroyed

Dynamic SIMS

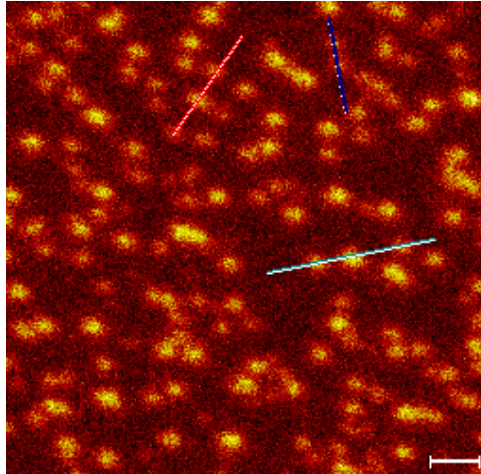


- Material removal
- Elemental analysis
- Depth profiling

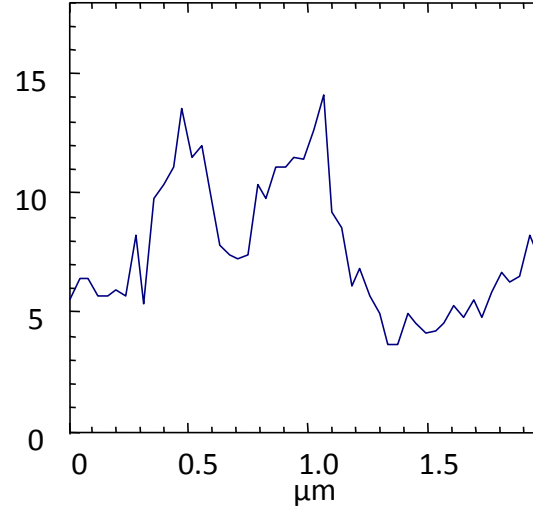
Courtesy Gregory L. Fisher, Physical Electronics



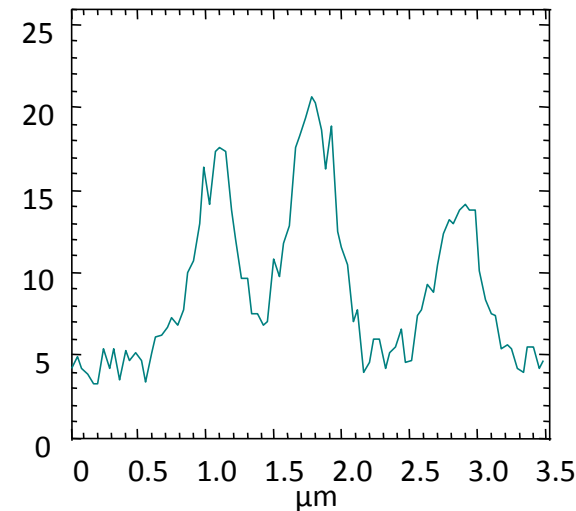
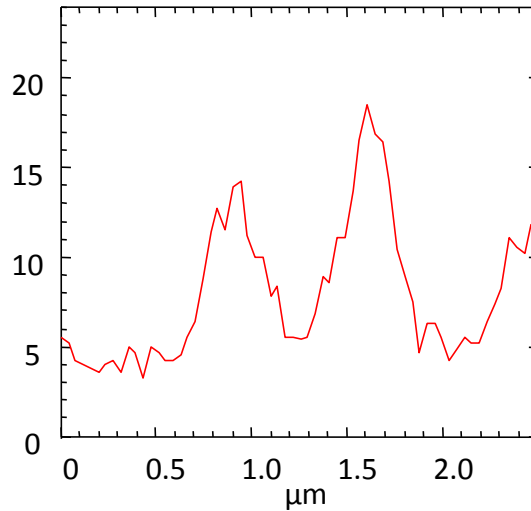
InAs/GaAs Quantum Dots

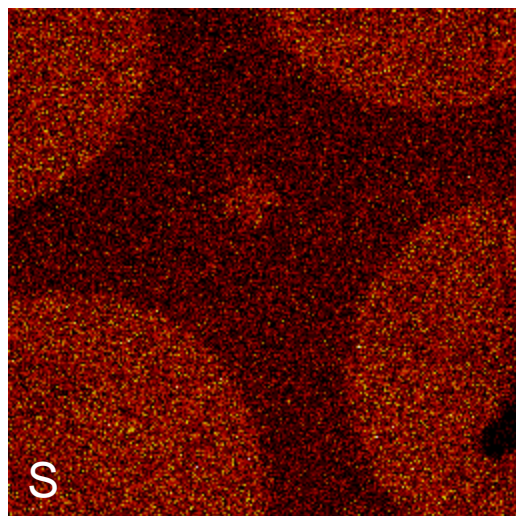
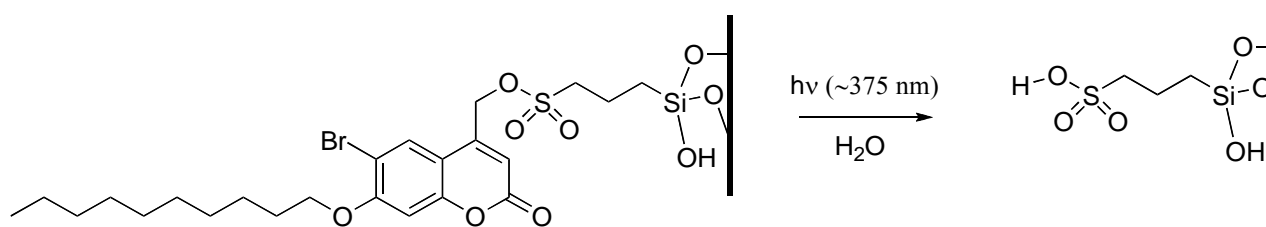


Cts: 550893; Max: 36; Scale: 1 μ m

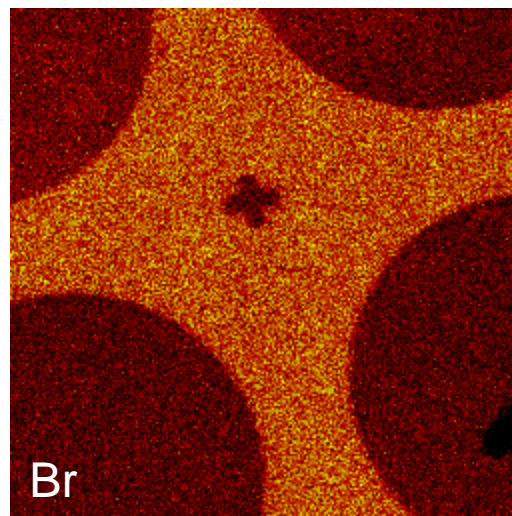


In⁺ Linescans of Quantum Dots

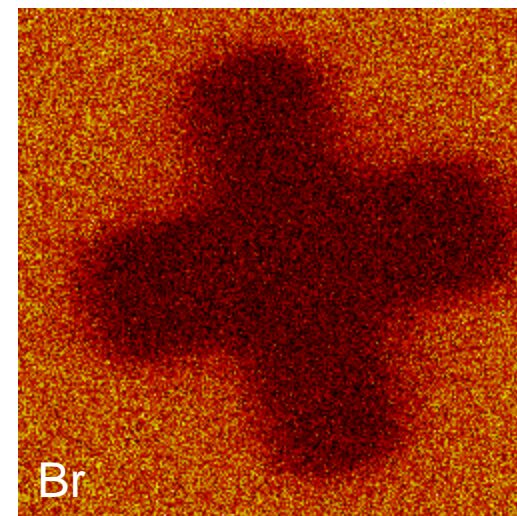




400 μm



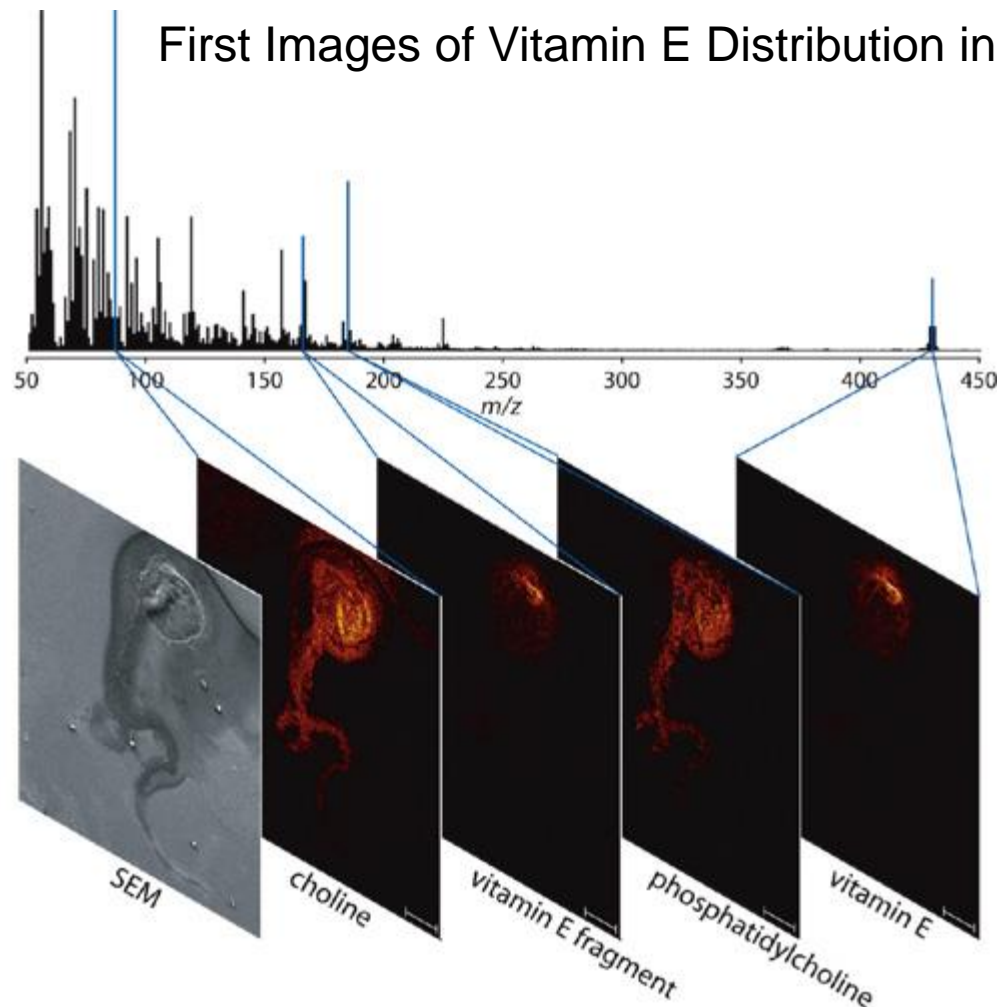
400 μm



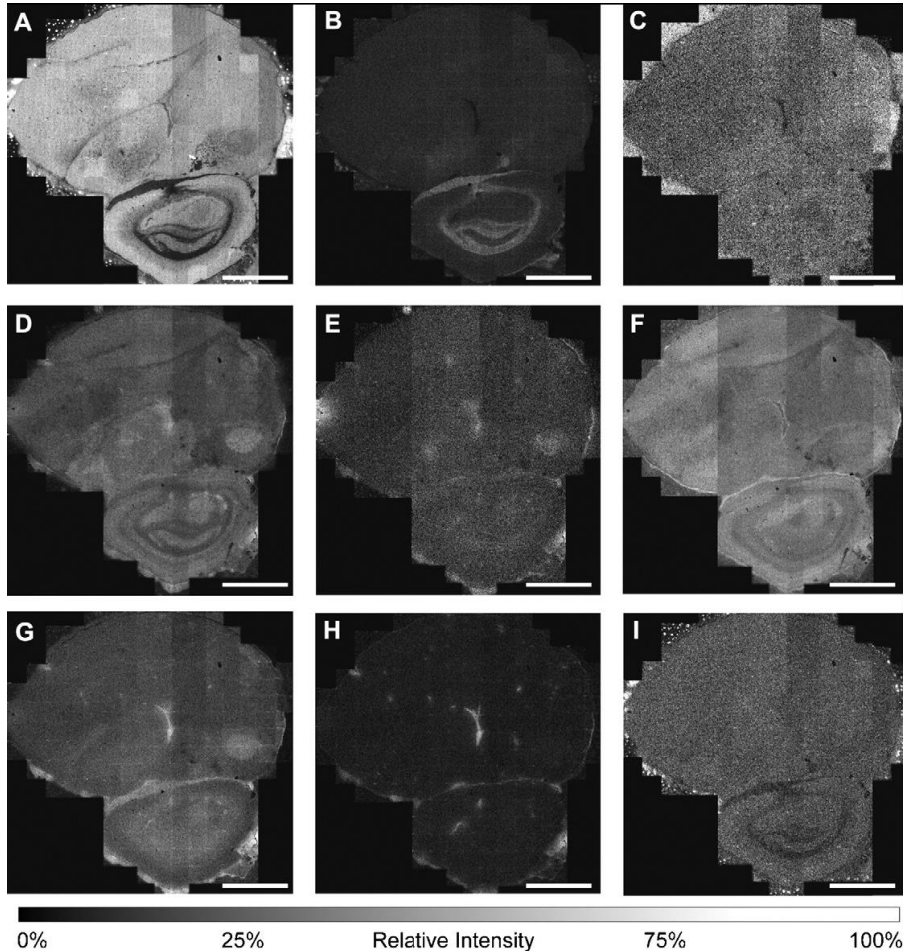
50 μm

Courtesy Josh Ritchey, Audrey Bowen, Ralph Nuzzo and Jeffrey Moore, University of Illinois

TOF-SIMS Ion Images of an Isolated Neuron

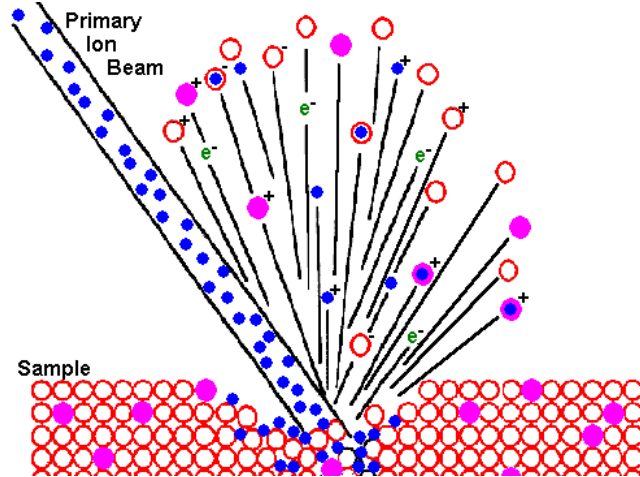


Courtesy E.B. Monroe,
J.C. Jurchen, S.S. Rubakhin,
J.V. Sweedler. University of
Illinois at Urbana-Champaign



Selected ion images from the songbird brain. Each ion image consists of ~11.5 million pixels within the tissue section and is the combination of 194 individual 600m×600m ion images prepared on the same relative intensity scale. Ion images are (A) phosphate PO_3^- (m/z 79.0); (B) cholesterol (m/z 385.4); (C) arachidonic acid $\text{C}_{20:4}$ (m/z 303.2); (D) palmitic acid $\text{C}_{16:0}$ (m/z 255.2); (E) palmitoleic acid $\text{C}_{16:1}$ (m/z 253.2); (F) stearic acid $\text{C}_{18:0}$ (m/z 283.3); (G) oleic acid $\text{C}_{18:1}$ (m/z 281.2); (H) linoleic acid $\text{C}_{18:2}$ (m/z 279.23); and (I) -linolenic acid $\text{C}_{18:3}$ (m/z 277.2). Scale bars = 2 mm.

Courtesy Kensey R. Amaya, Eric B. Monroe, Jonathan V. Sweedler, David F. Clayton.
International Journal of Mass Spectrometry **260**, 121 (2007).



In SIMS, the yield of secondary ions is strongly influenced by the electronic state of the material being analyzed.

$$I_s^m = I_p y_m \alpha^+ \theta_m \eta$$

I_s^m = secondary ion current of species m

I_p = primary particle flux

y_m = sputter yield

α^+ = ionization probability to positive ions

θ_m = fractional concentration of m in the layer

η = transmission of the analysis system



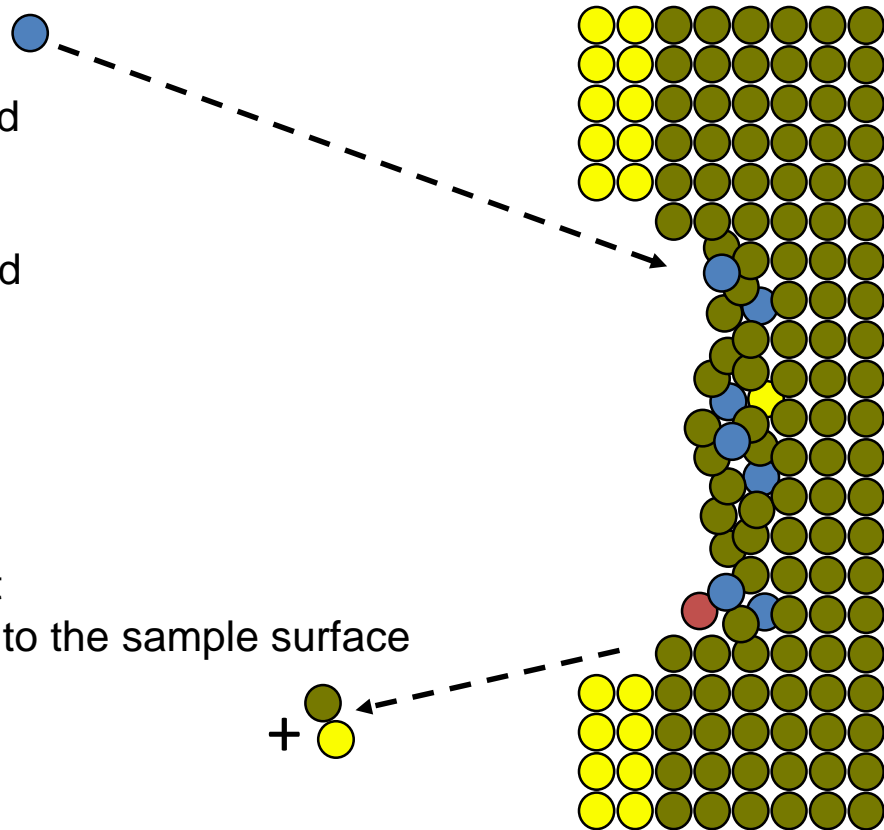
Total Ion Sputtering Yield

Sputter yield: ratio of number of atoms sputtered to number of impinging ions, typically 5-15

Ion sputter yield: ratio of ionized atoms sputtered to number of impinging ions, 10^{-6} to 10^{-2}

Ion sputter yield may be influenced by:

- Matrix effects
- Surface coverage of reactive elements
- Background pressure in the sample environment
- Orientation of crystallographic axes with respect to the sample surface
- Angle of emission of detected secondary ions



First principles prediction of ion sputter yields is not possible with this technique.

Courtesy of Prof. Rockett



Effect of Primary Beam on Secondary Ion Yields

H																		He
Li	Be											B	C	N	O	F		Ne
Na	Mg											Al	Si	P	S	Cl		Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br		Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I		Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At		Rn
Fr	Ra	Ac																

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Graphics courtesy of Charles Evans & Associates web site <http://www.cea.com>

Oxygen bombardment

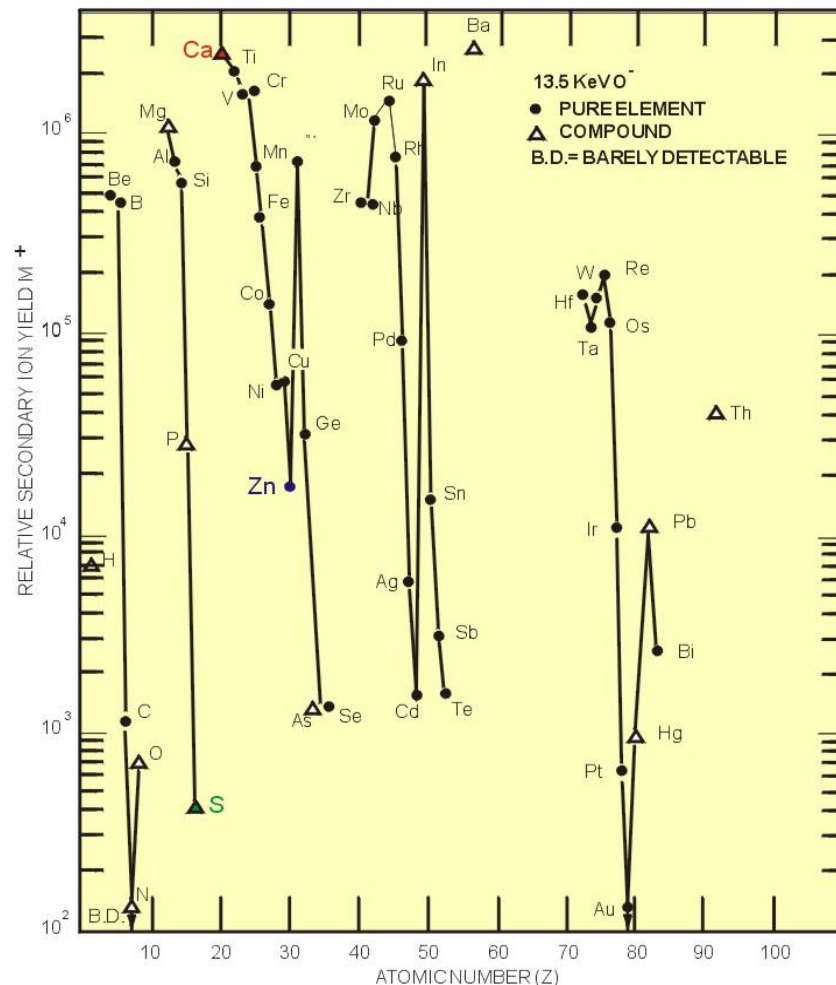
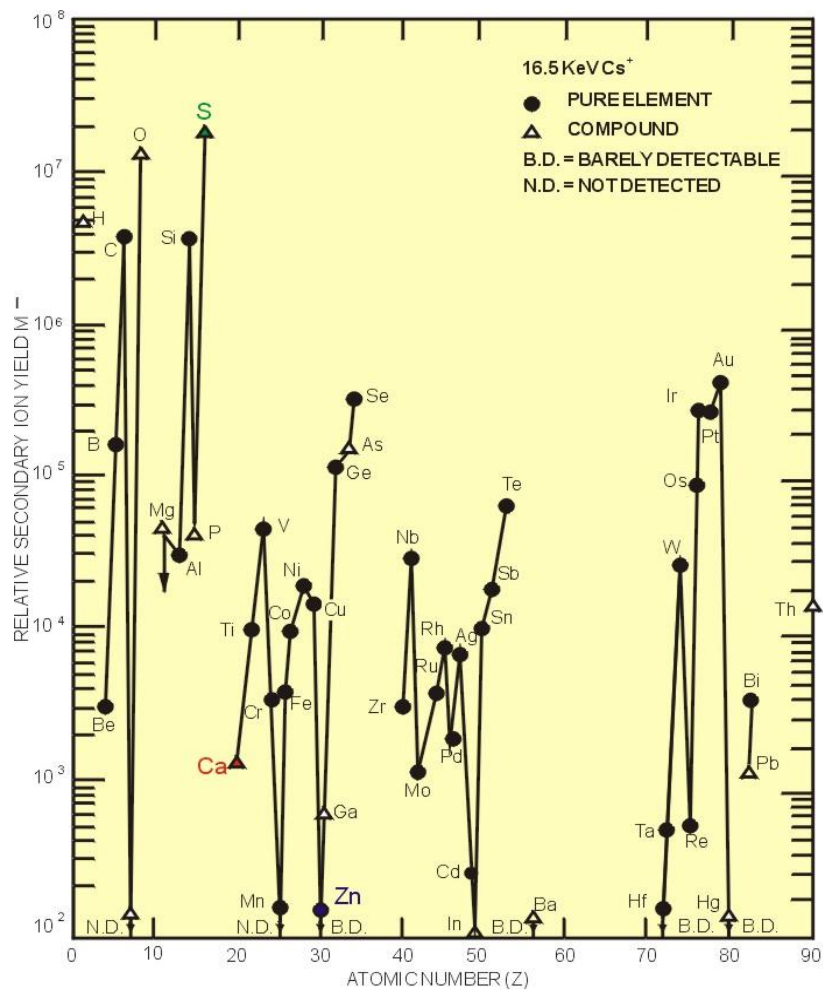
When sputtering with an oxygen beam, the concentration of oxygen increases in the surface layer and metal-oxygen bonds are present in an oxygen-rich zone. When the bonds break during the bombardment, secondary ion emission process, oxygen becomes negatively charged because of its high electron affinity and the metal is left with the positive charge. Elements in yellow analyzed with oxygen bombardment, positive secondary ions for best sensitivity.

Cesium bombardment

When sputtering with a cesium beam, cesium is implanted into the sample surface which reduces the work function allowing more secondary electrons to be excited over the surface potential barrier. With the increased availability of electrons, there is more negative ion formation. Elements in green analyzed with cesium, negative secondary ions for best sensitivity.



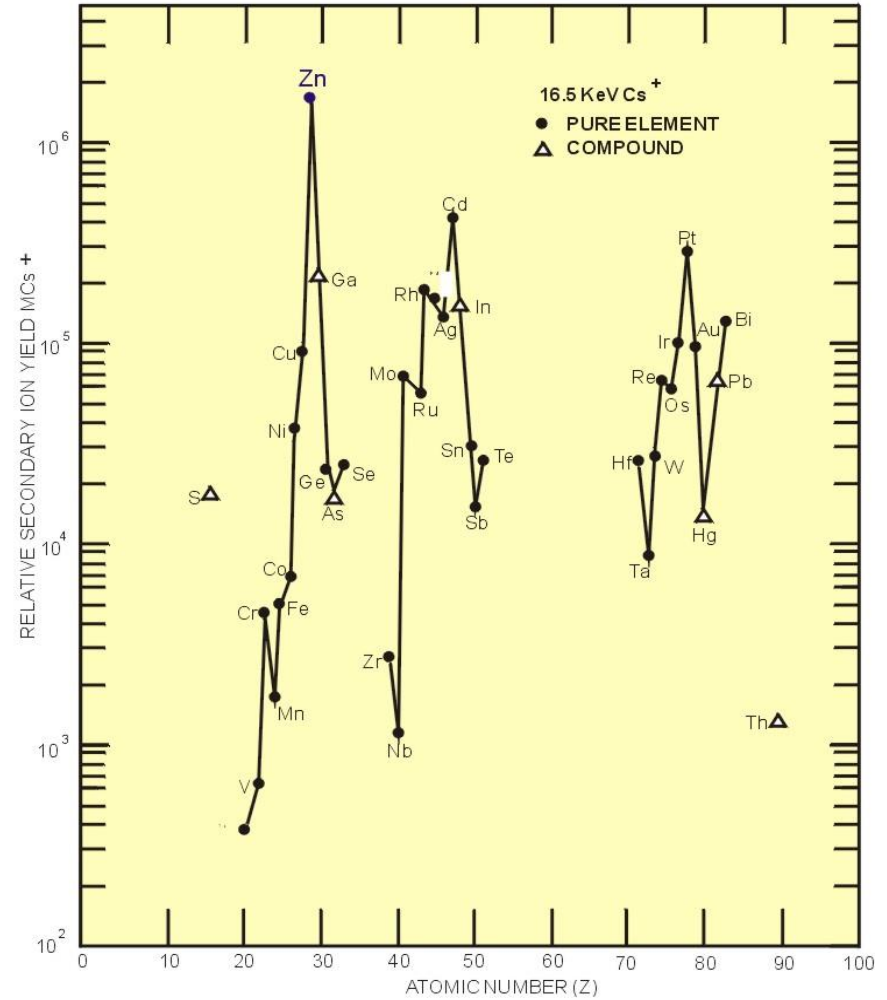
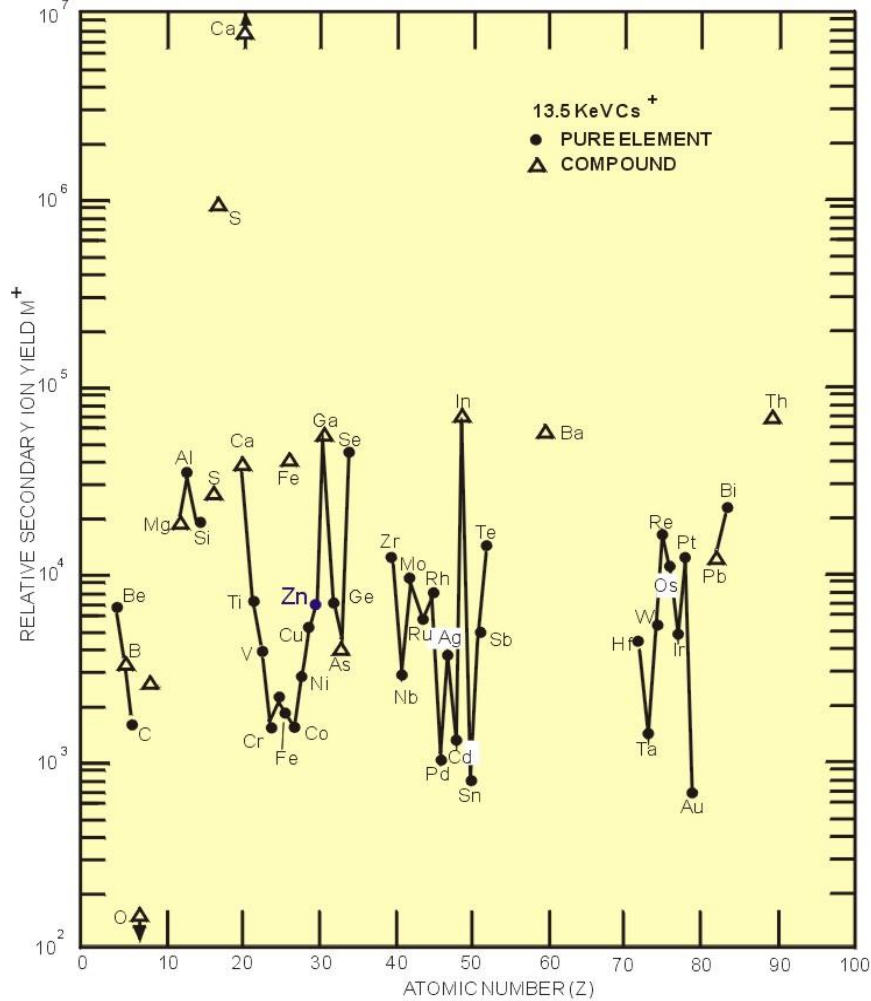
Relative Secondary Ion Yield Comparison



From Storms, et al., Anal. Chem. 49, 2029 (1977).



Relative Secondary Ion Yield Comparison

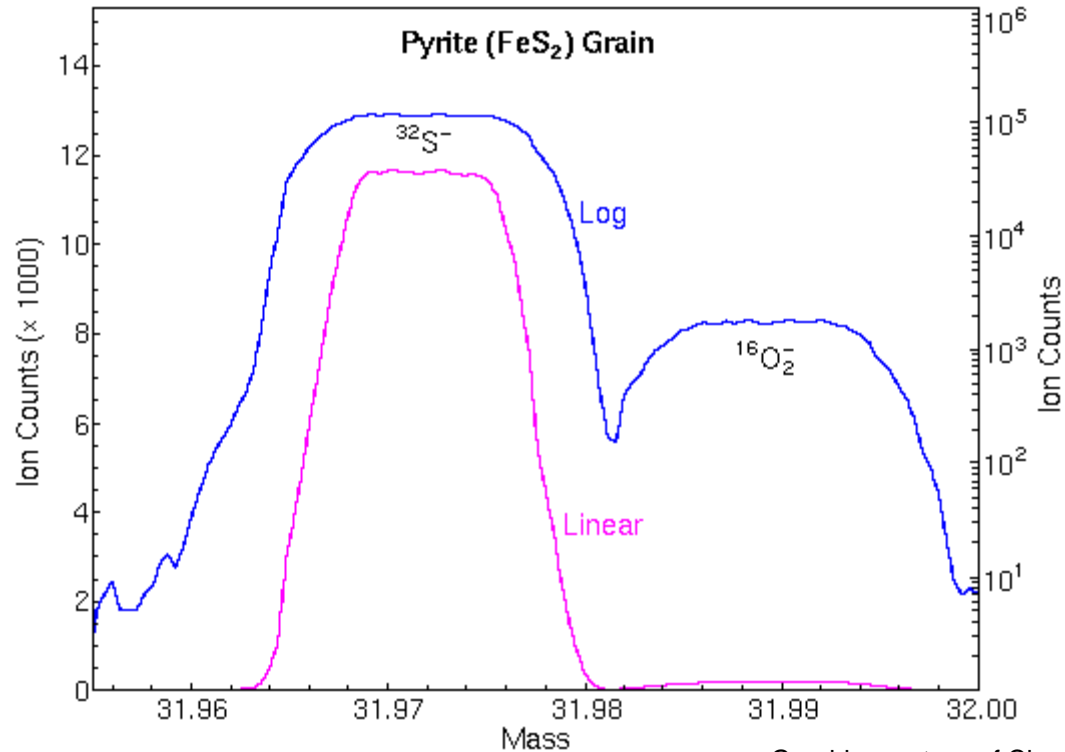


From Storms, et al., Anal. Chem. 49, 2023 (1977).



Definition of Mass Resolution

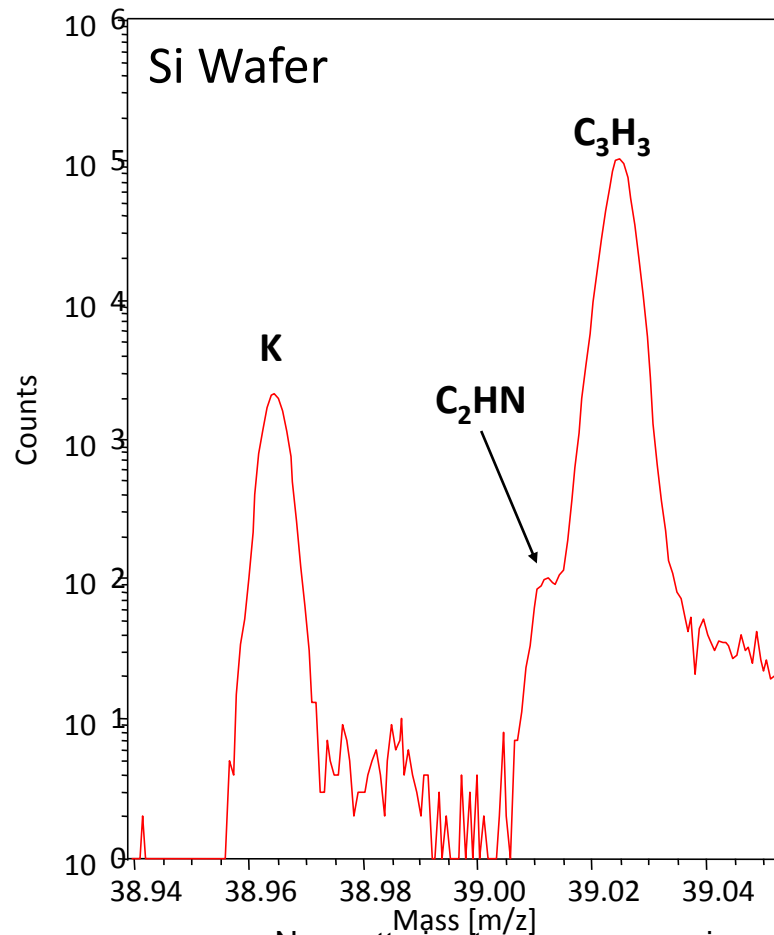
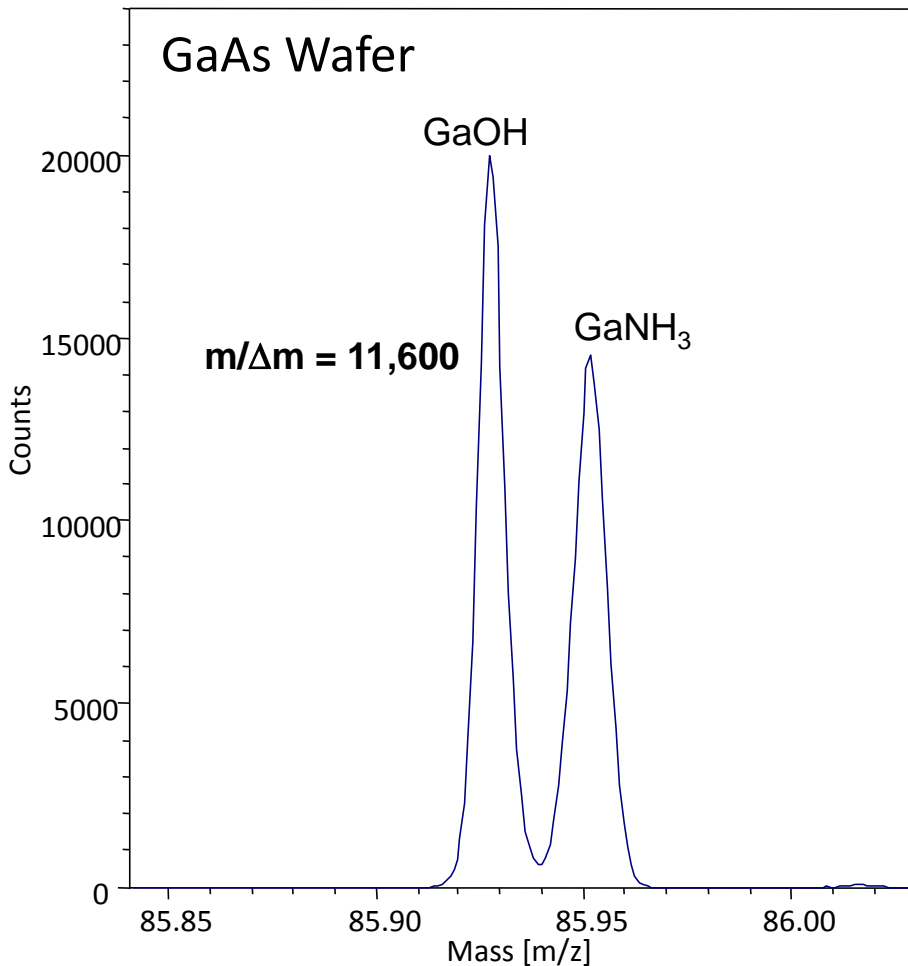
Mass resolution defined by $m/\Delta m$
Mass resolution of ~ 1600 required to resolve ^{32}S from $^{16}\text{O}_2$



Graphic courtesy of Charles Evans & Associates web site
<http://www.cea.com>



Trace Analysis



No sputtering to remove organics on surface.
Large C₃H₃ peak does not have a tail to lower mass which would obscure C₂HN and K⁺



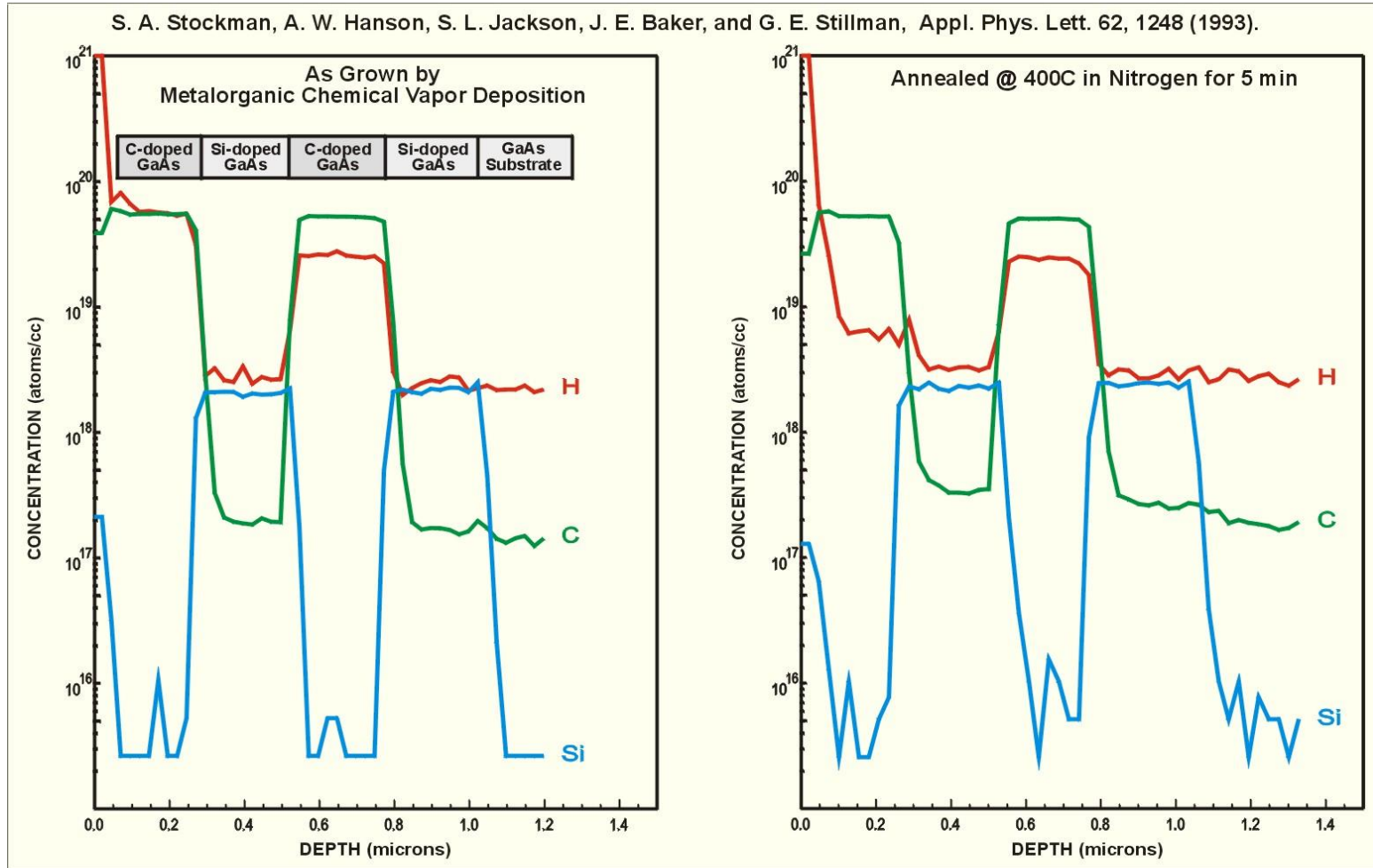
Comparison of Static and Dynamic SIMS

TECHNIQUE	STATIC	DYNAMIC
FLUX	$< 10^{13}$ ions/cm ² (per experiment)	$\sim 10^{17}$ ions/cm ² (minimum dose density)
INFORMATION	Elemental + Molecular	Elemental
SENSITIVITY	1 ppm	< 1 ppm (ppb for some elements)
TYPE OF ANALYSIS	Surface Mass Spectrum 2D Surface Ion Image	Depth Profile Mass Spectrum 3D Image Depth Profile
SAMPLING DEPTH	2 monolayers	10 monolayers
SPATIAL RESOLUTION	0.1 – 1.0 μm	0.1 -1.0 μm
SAMPLE DAMAGE	Minimal	Destructive in analyzed area – up to 500 μm per area



Depth Profile Application with Hydrogen

S. A. Stockman, A. W. Hanson, S. L. Jackson, J. E. Baker, and G. E. Stillman, Appl. Phys. Lett. 62, 1248 (1993).

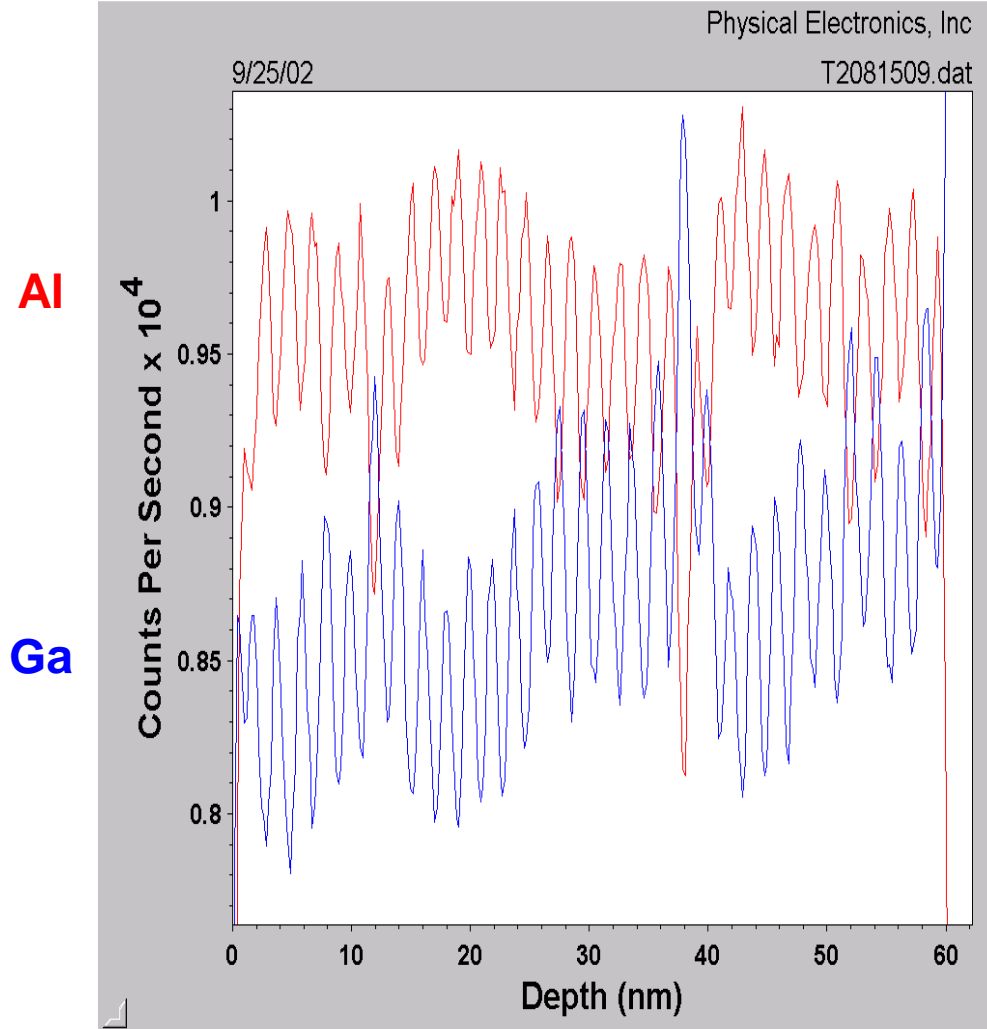


Detects hydrogen

Large dynamic range



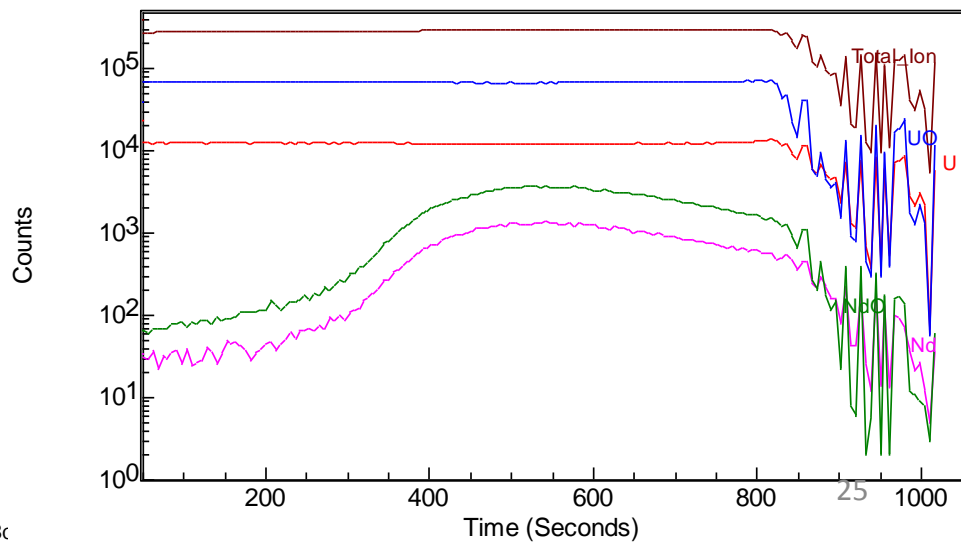
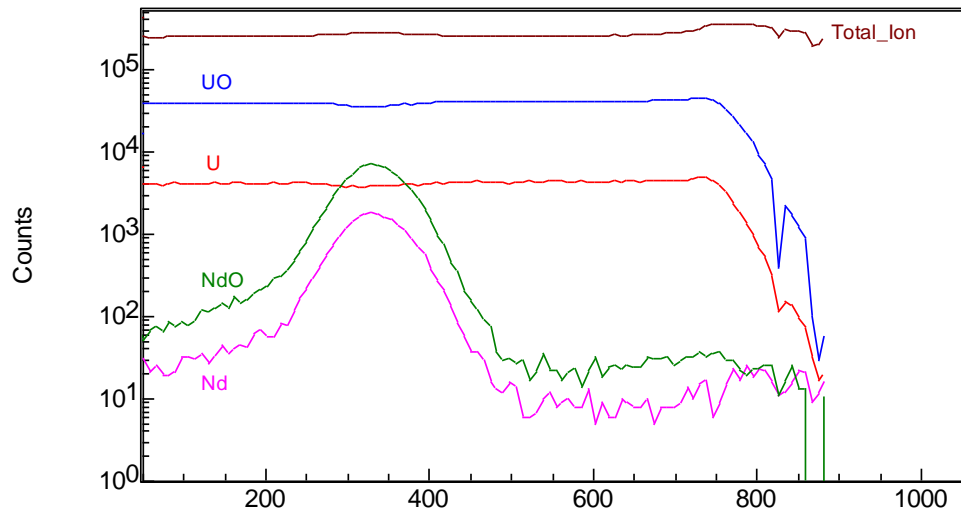
GaAs/AlGaAs Depth Profile



Analysis beam: 15kV Ga⁺
Sputter Beam: 300V O₂⁺
with oxygen flood



Depth Profile Beam Alignment



$$I_s^m = I_p y_m \alpha^+ \theta_m \eta$$

Level Profile:

$$RSF = \frac{I_m}{I_i} \rho_i$$

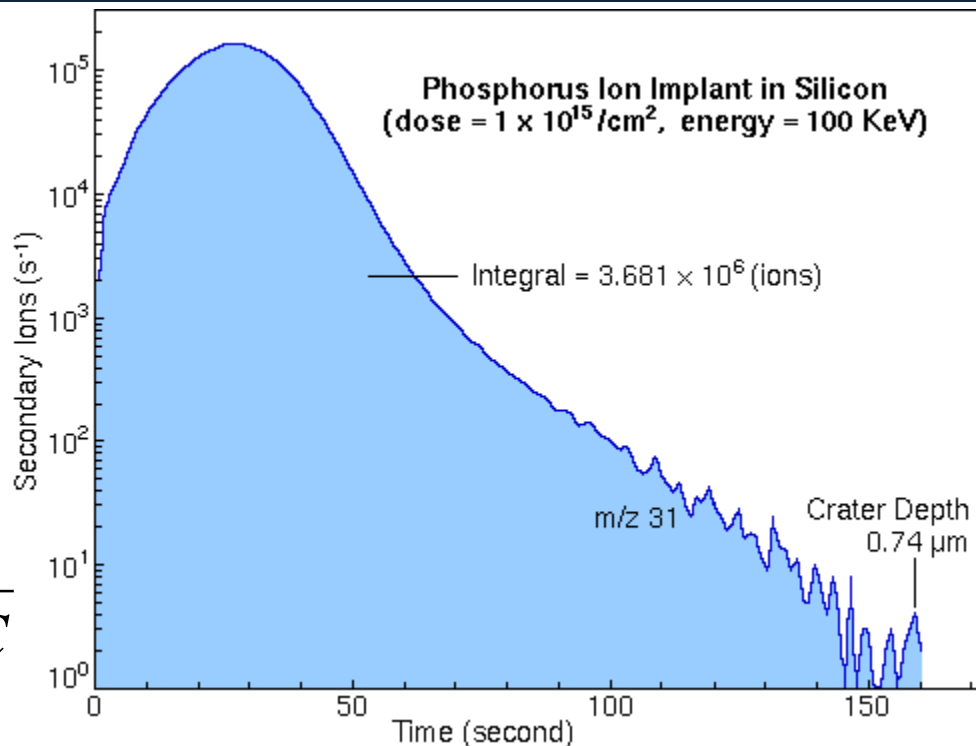
Gaussian Profile:

$$RSF = \frac{I_m \phi C t}{d \sum I_i - d I_b C}$$

Where:

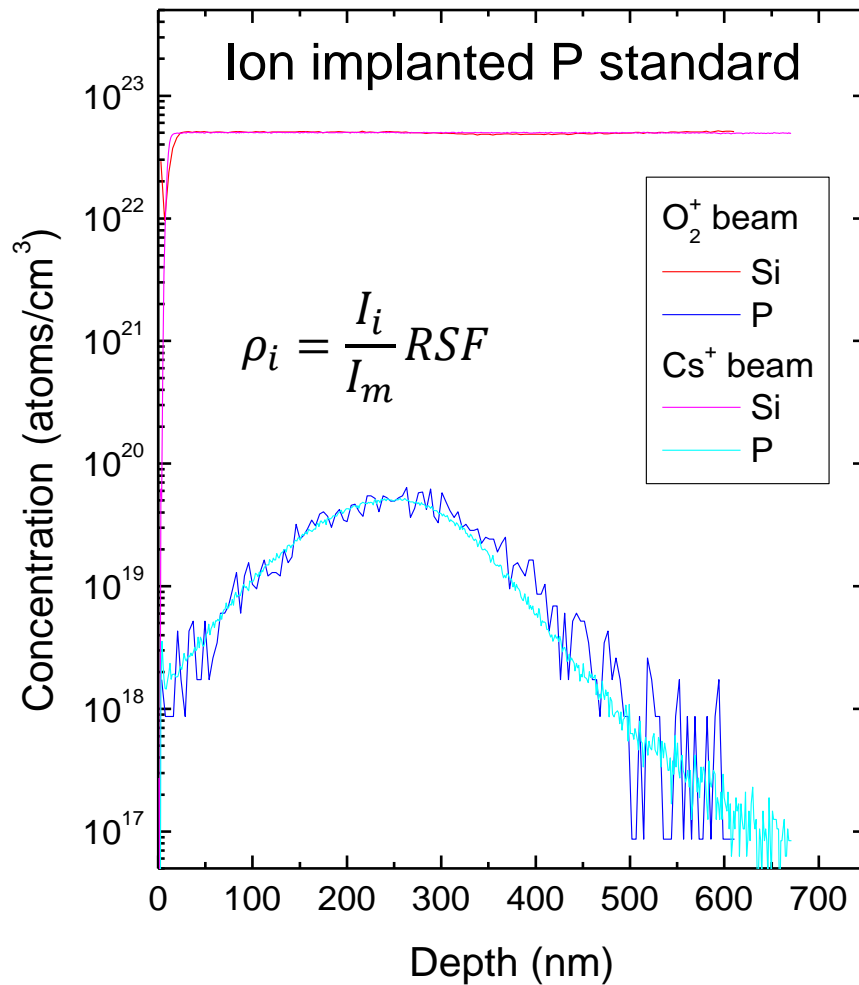
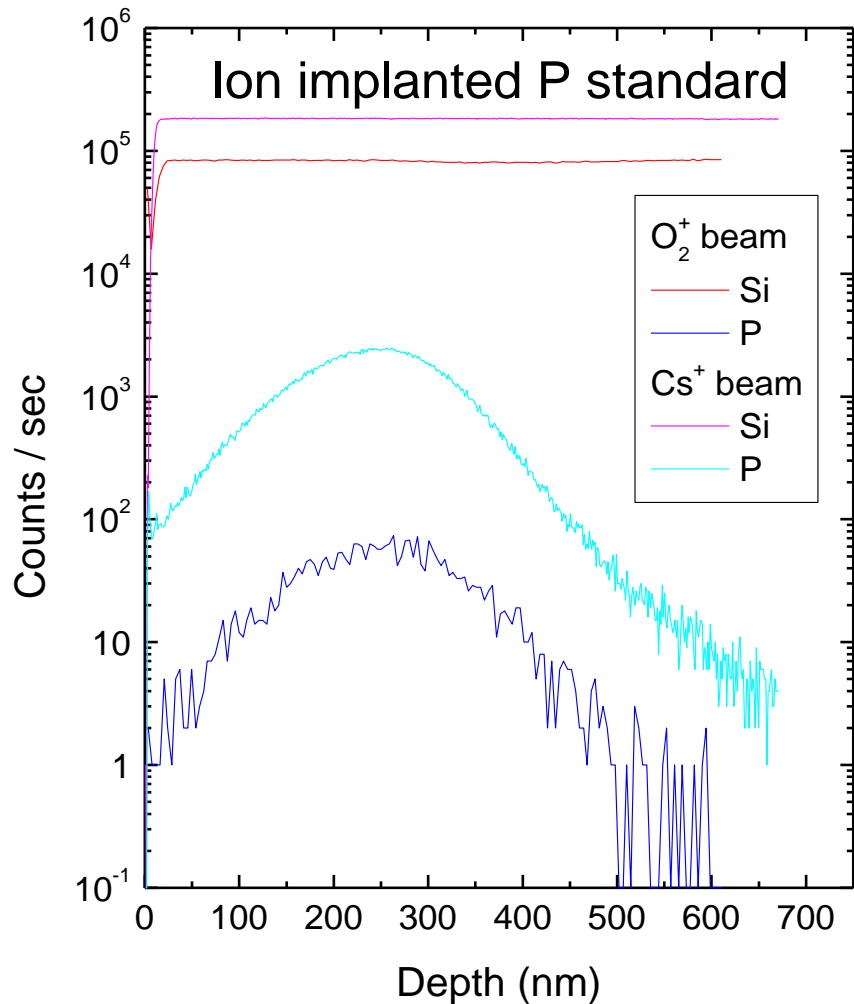
RSF = Relative Sensitivity Factor
 I_m, I_i = ion intensity (counts/sec)
 ρ = atom density (atoms/cm³)
 ϕ = implant fluence (atoms/cm²)

C = # measurement cycles
 t = analysis time (s/cycle)
 d = crater depth (cm)
 I_b = background ion counts



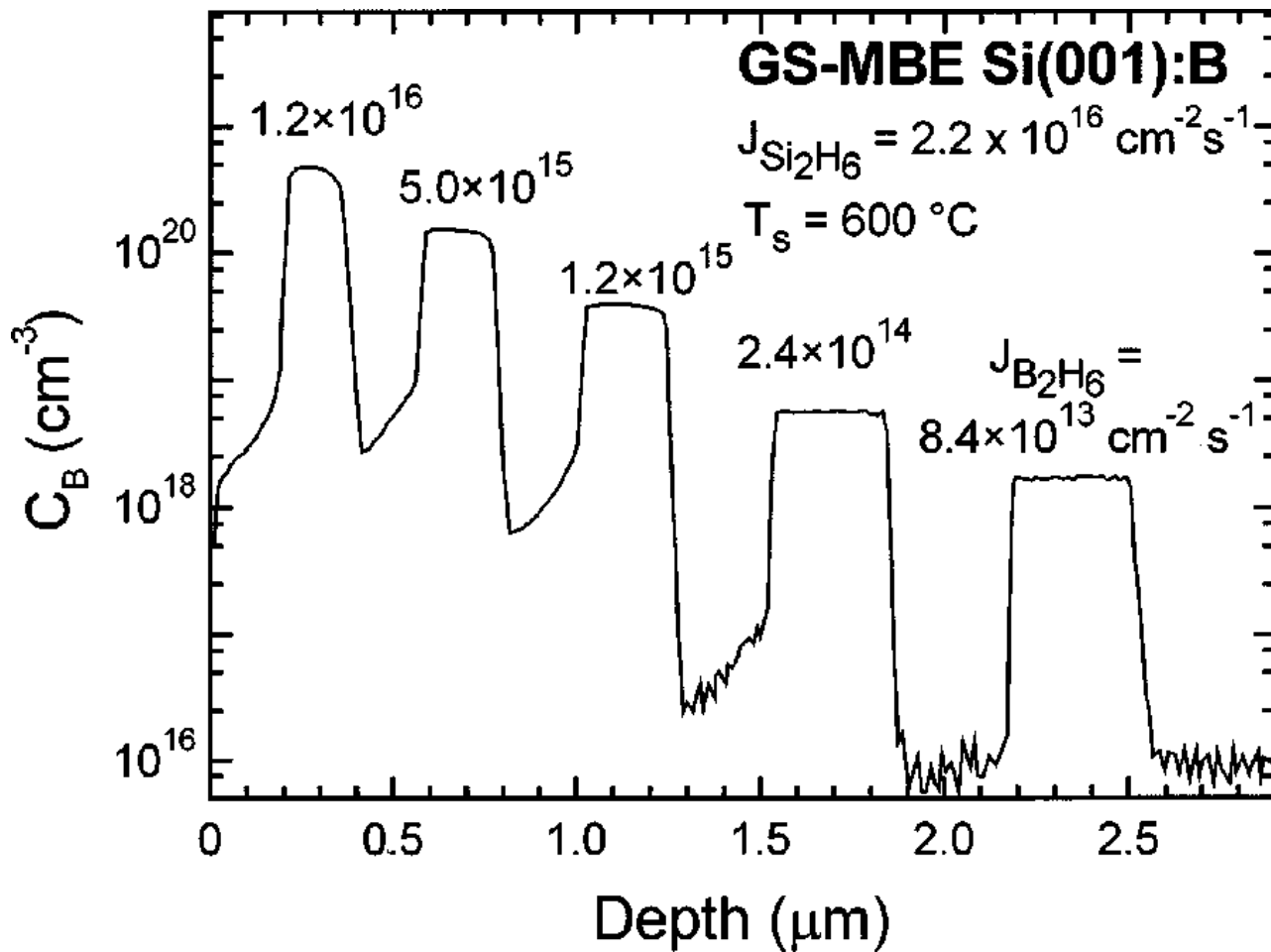


Positive and Negative Secondary Ions



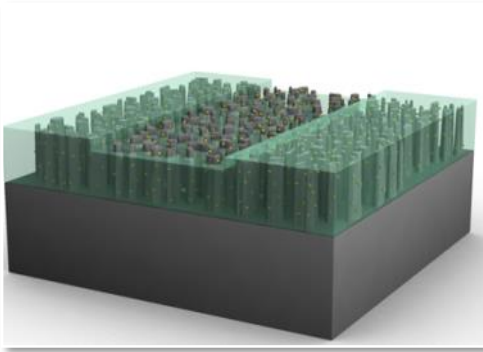


B Depth Profile in Si(001)

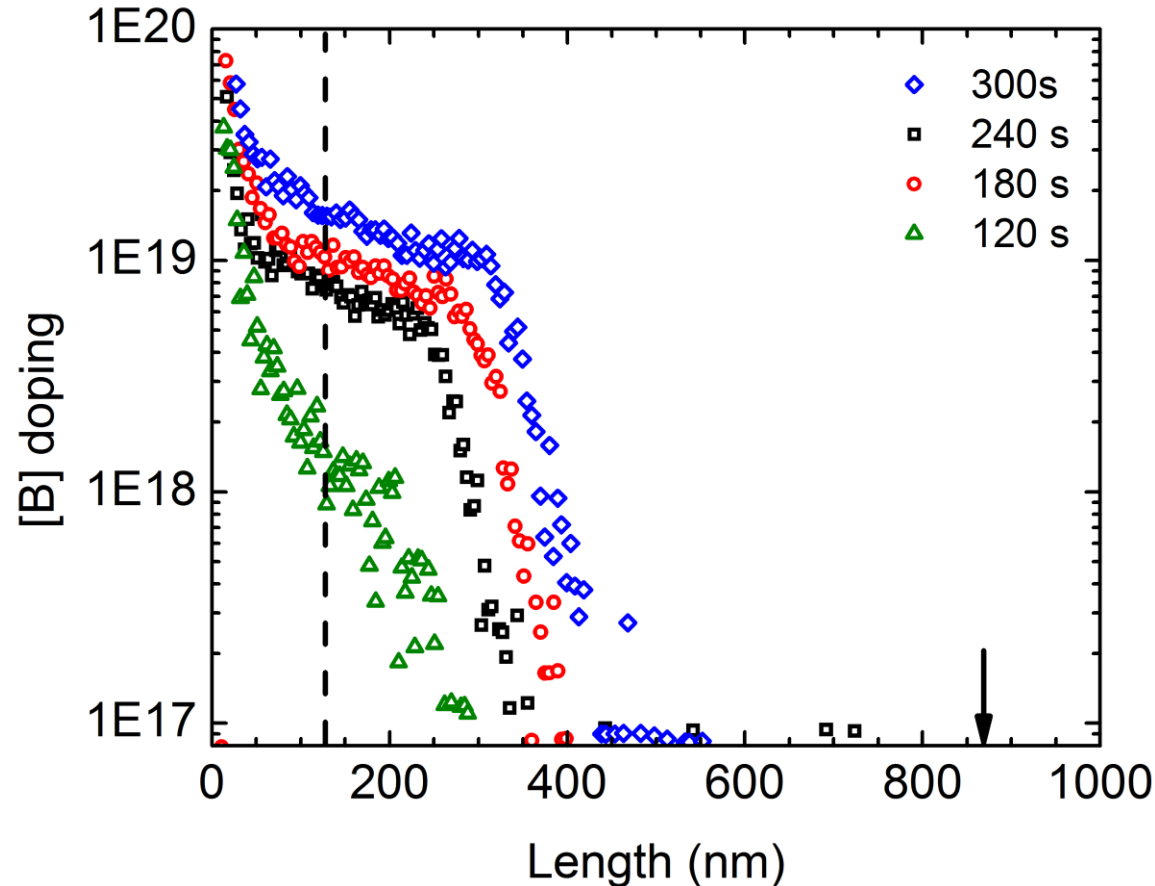
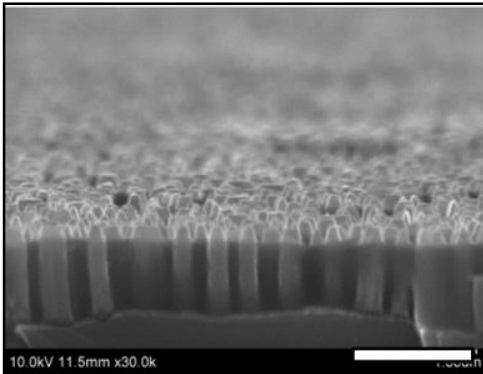


SIMS depth profiles through a B modulation-doped Si(001):B film grown by GS-MBE from Si_2H_6 and B_2H_6 at $T_s=600 \text{ }^\circ\text{C}$. The incident Si_2H_6 flux was $J_{\text{Si}_2\text{H}_6} = 2.2 \times 10^{16} \text{ cm}^{-2} \text{ s}^{-1}$ while the B flux $J_{\text{B}_2\text{H}_6}$ was varied from 8.4×10^{13} to $1.2 \times 10^{16} \text{ cm}^{-2} \text{ s}^{-1}$. The deposition time for each layer was constant at 1 h.

G. Glass, H. Kim, P. Desjardins, N. Taylor, T. Spila, Q. Lu, and J. E. Greene. Phys. Rev. B, **61**,7628 (2000).



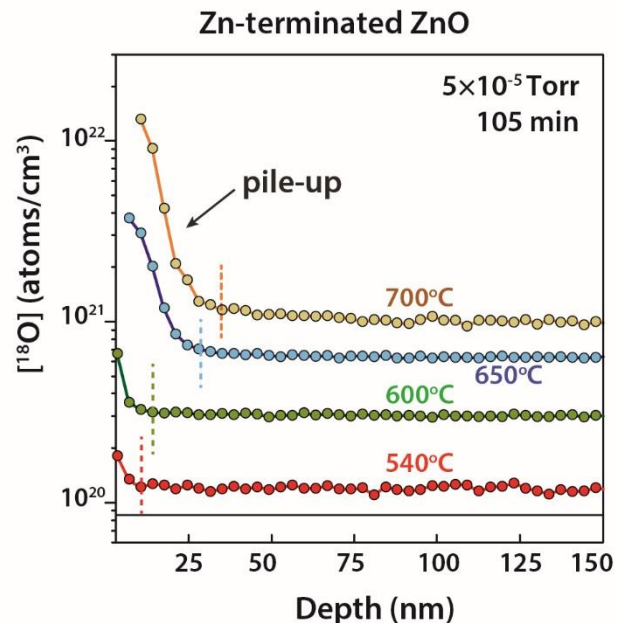
Dope NW tips by SODs



J.S Sadhu, H. Tian, T. Spila, J. Kim, B. Azeredo, P. Ferreira, and S. Sinha. *Nanotechnology* **25**, 375701 (2014).

Charged point defects interact with space charge in the near-surface region via 3 mechanisms

1. Field-induced drift (O in TiO_2)¹
2. Change in charge state of defect with local potential (B in Si)²
3. Potential energy-dependent formation energy of V_O (O in SrTiO_3)³



¹P. Gorai et al., *Appl. Phys. Lett.* **103**, 141601 (2013).

³R.A. De Souza and M. Martin, *PCCP* **10**, 2356 (2008).

^{18}O piles up in the first 10-30 nm in ZnO and TiO_2 ¹

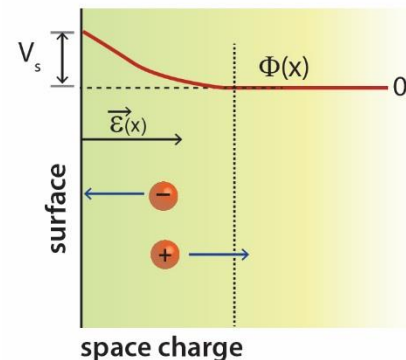
- Amount of pile-up (P) = integrated area between pile-up and bulk extrapolated profiles

Analytical model quantifies effects⁴

- Drift opposite to diffusion causes pile-up
- Drift in diffusion direction depletes near-surface of mobile defects
- P increases linearly with time & flux, quadratically with V_s
- V_s of only a few meV can cause the amount of pile-up observed

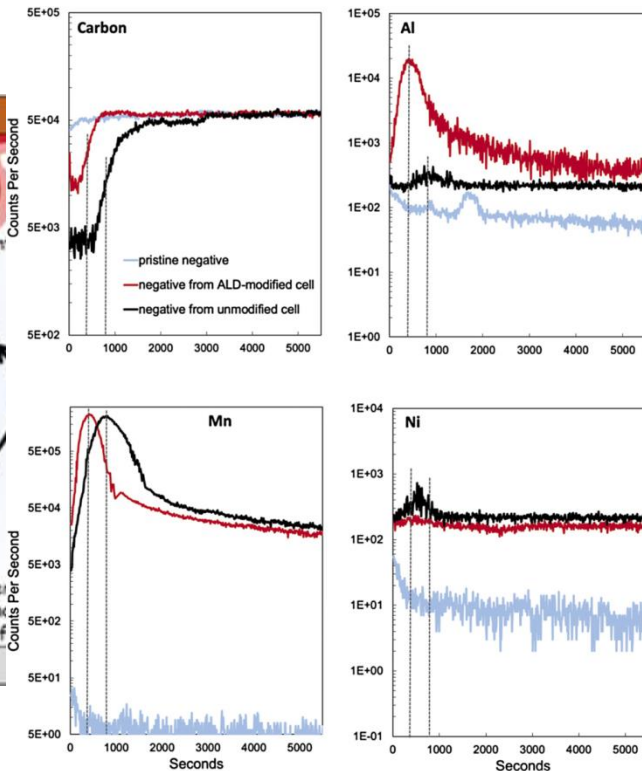
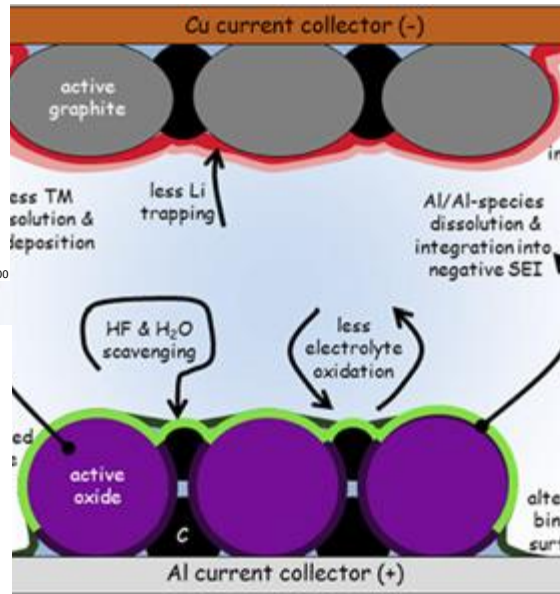
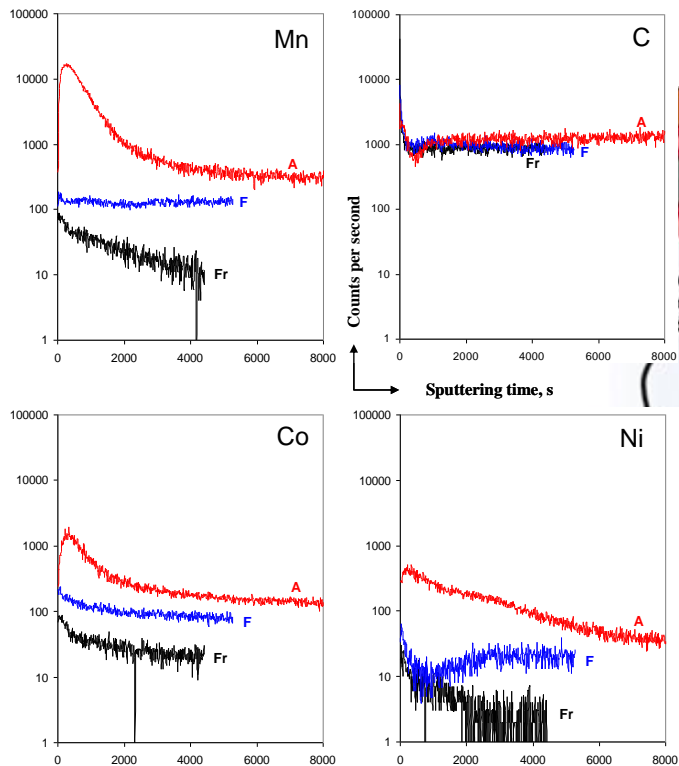
²P. Gorai et al., *J. Appl. Phys.* **111**, 094510 (2012).

⁴P. Gorai and E. G. Seebauer, *Appl. Phys. Lett.* **105**, 021604 (2014).



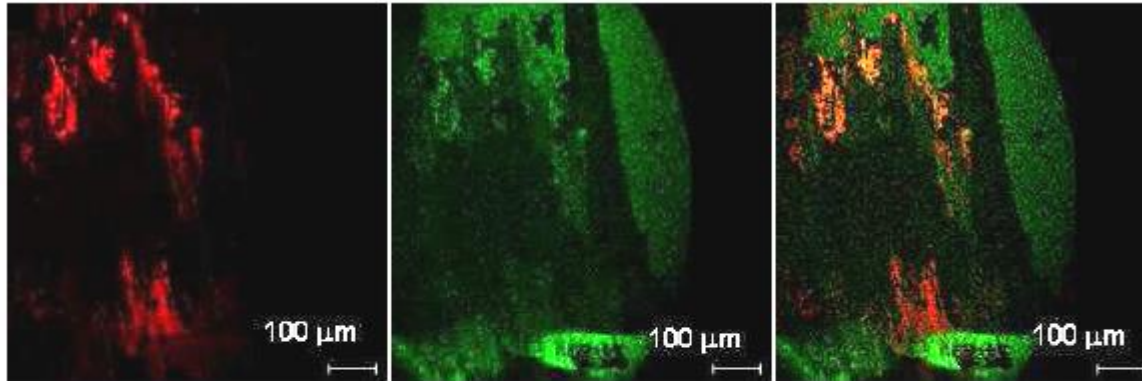


Transition-Metal Accumulation on Anodes in Li-ion Batteries

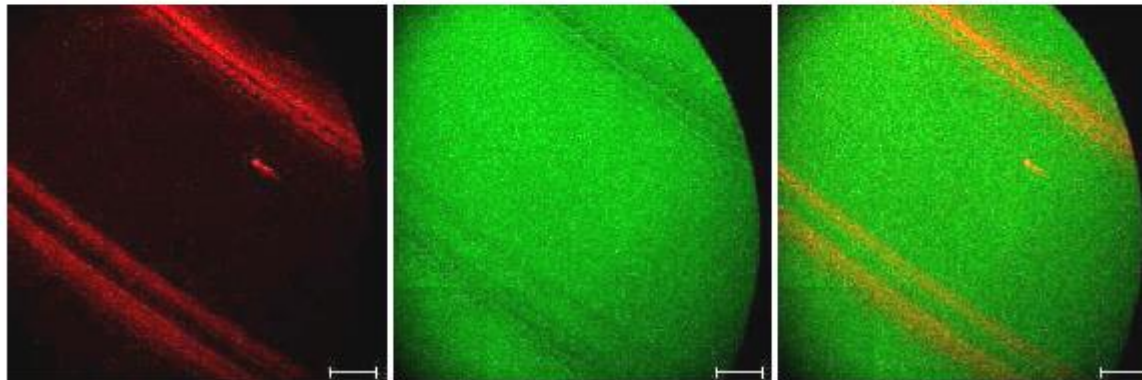




Diamond-Like-Carbon Friction Testing



DLC coated ball



DLC coated disk

Oxygen

Carbon

C + O
Overlay

wear tracks and scars formed on DLC-coated disk and ball sides during test in dry oxygen

Courtesy O.L. Eryilmaz and A. Erdemir
Energy Systems Division,
Argonne National Laboratory
Argonne, IL 60439 USA

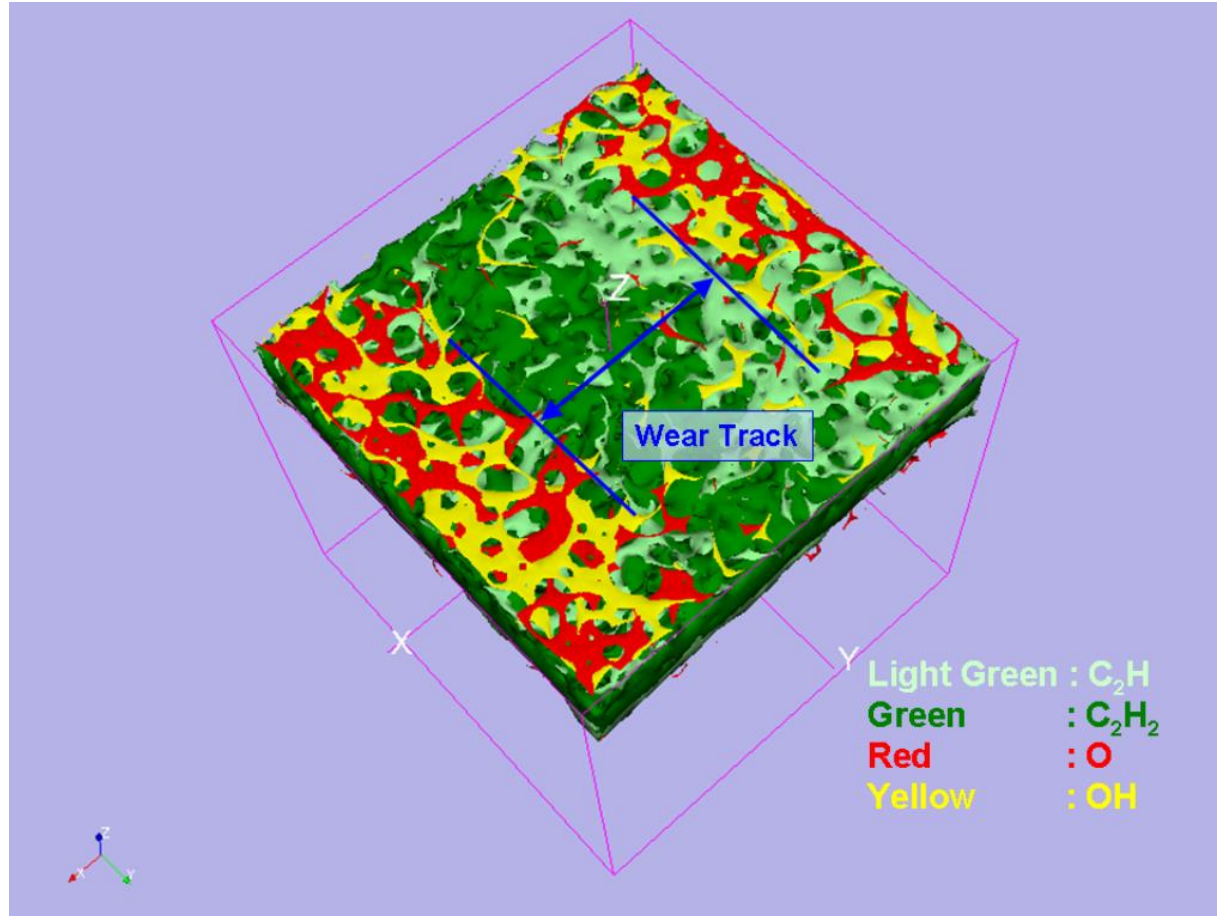


3-D TOF-SIMS imaging of DLC

Wear track from hydrogenated DLC tested in dry nitrogen

Courtesy O.L. Eryilmaz and A. Erdemir

Energy Systems Division, Argonne National Laboratory Argonne, IL 60439 USA



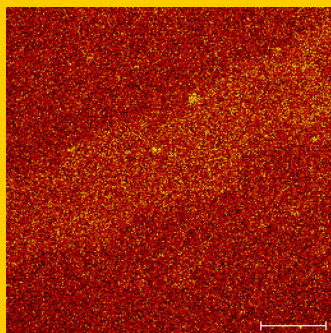


3-D TOF-SIMS Movies of DLC

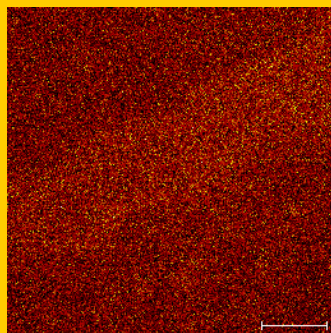
NFC6 H2 Environment TOF-SIMS Images

Courtesy O.L. Eryilmaz and A. Erdemir

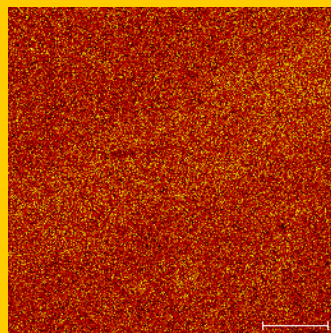
Energy Systems Division, Argonne National Laboratory Argonne, IL 60439 USA



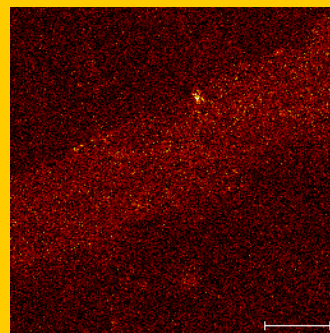
H



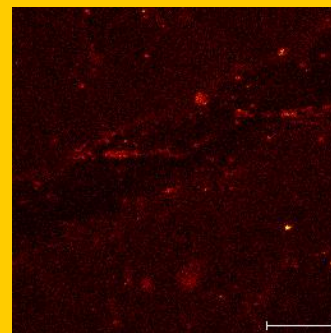
CH



C₂H

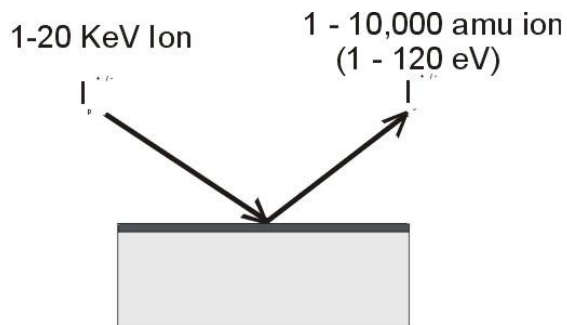


C₂H₂



O

Probe/Detected Species



Information

Surface Mass Spectrum
2D Surface Ion Image
Elemental Depth Profiling
3D Image Depth Profiling

Elements Detectable

H and above

Sensitivity

ppb - atomic %

Analysis Diameter/Sampling Depth

~1 μm - several mm/0.5 - 1nm



Where do Drug Molecules go Inside of Cells? A New Method to Probe the Composition of Cellular Organelles

Ashley Ellsworth¹, Corryn E. Chini², Ben Johnson³,
Michael M. Tamkun³, Gregory L. Fisher¹, and Mary L. Kraft²

¹ Physical Electronics, Chanhassen, MN, USA

² University of Illinois at Urbana-Champaign, School of Chemical Sciences, Urbana, IL, USA

³ Colorado State University, Department of Biomedical Science, Fort Collins, CO, USA





Atom Probe Tomography for Additive Manufacturing

Katherine P. Rice, Yimeng Chen Ty J. Prosa, Robert M. Ulfing

CAMECA Instruments, Inc. 5470 Nobel Drive, Madison, WI USA



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